



Lichen bioindicators for mercury contamination

*Rotary Kiln,
New Almaden Quicksilver County Park*

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Outline

- Utility of lichen as a bioindicator of atmospheric mercury deposition.
- Need for monitoring with lichen to identify deposition hotspots.
- Recent results in Hg-mine impacted areas.
- Calibrating lichen Hg with direct measurements of Hg dry deposition.
- Application of method to the Carson River Superfund site.





Flavopunctelia sp.



Ramalina farinacea



Evernia prunastri



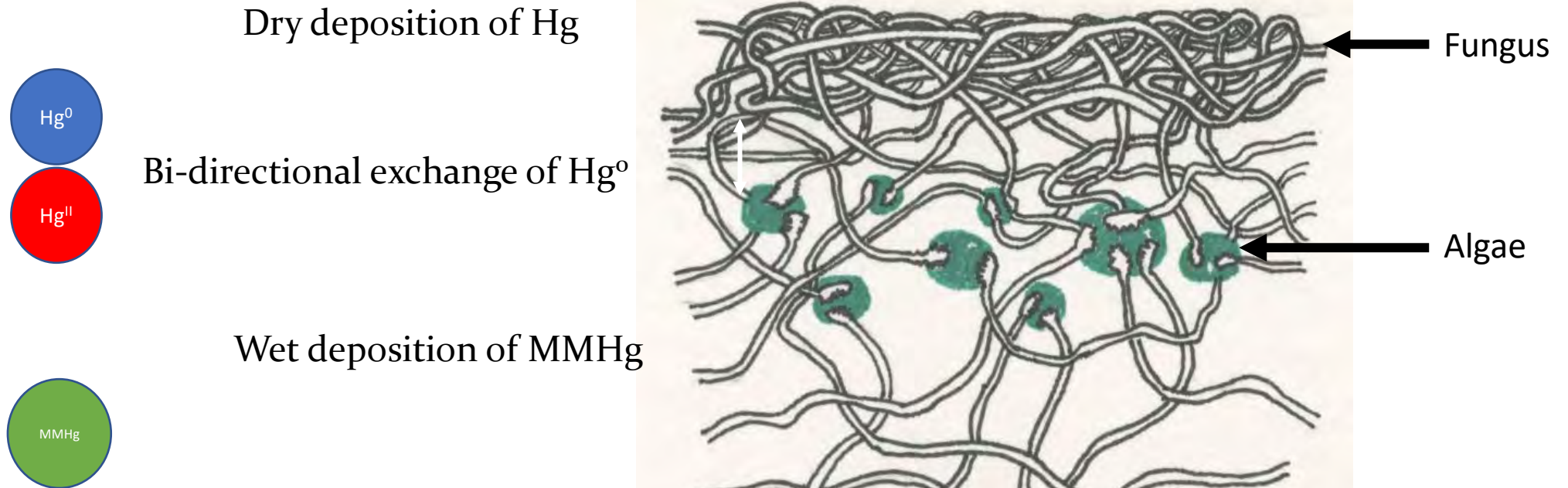
Usnea sp.



Ramalina menziesii

5 Principal Species of Lichen
in California Coast Range

How lichen interacts with mercury in air

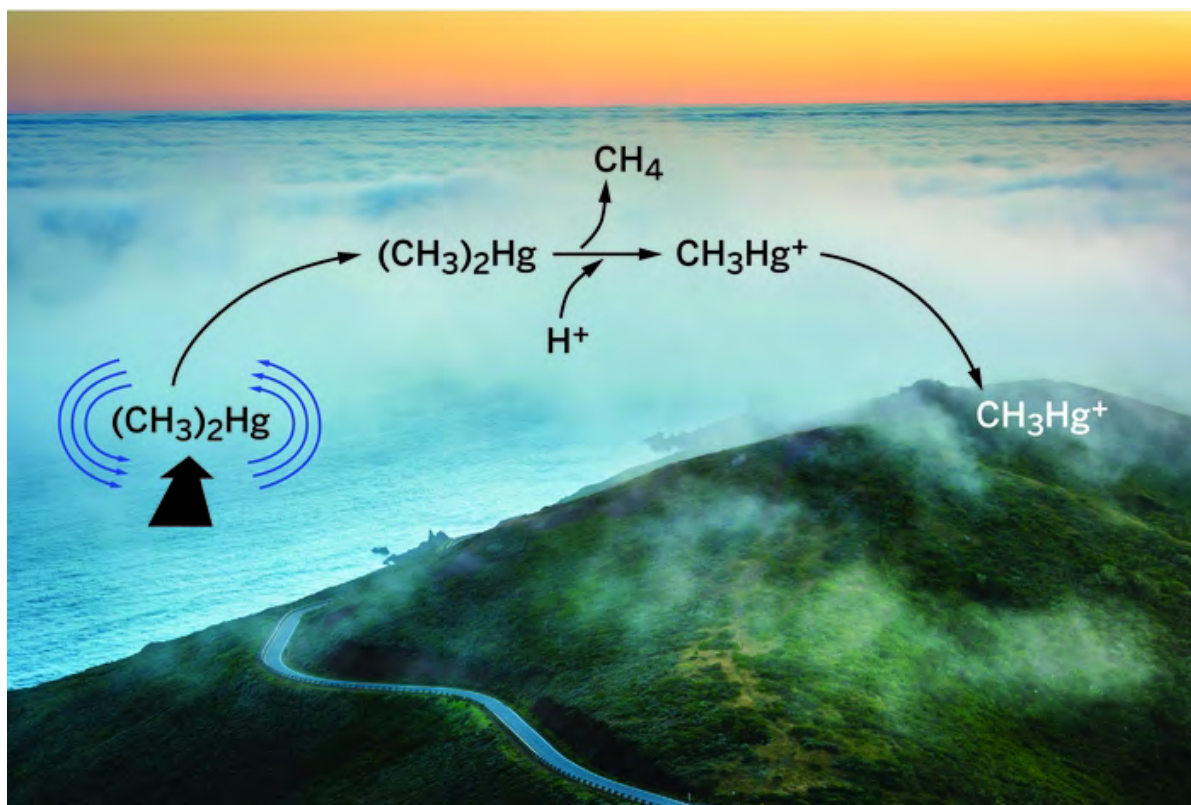


Fun Facts about Lichen and Hg

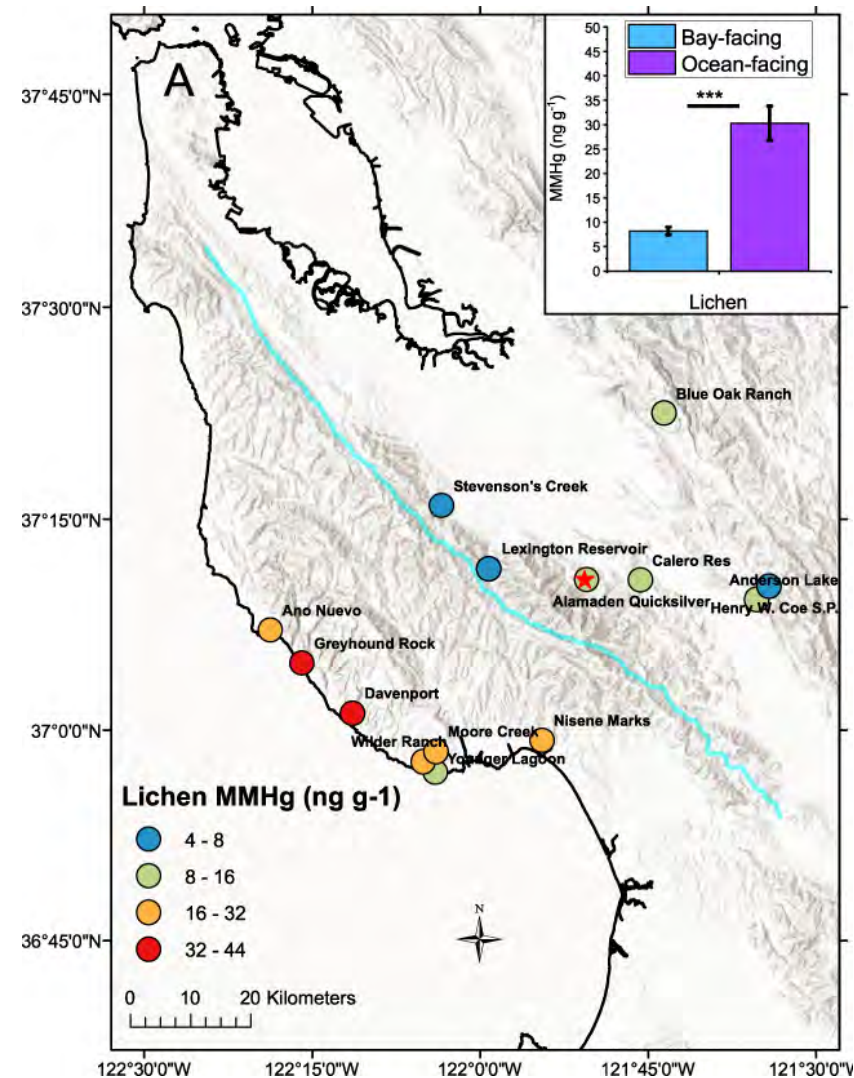
- Rapid uptake of Hg, especially Hg^{II}
- Residence time of Hg of ~5 years (Nieboer and Richardson, 1981)
- Time-integrated Hg accumulation rate reflects temporal variations in atmospheric Hg loads (Wittig, 1993)

Previous Work:

Methylmercury in *Ramalina menziesii* along a coastal fog gradient



C&EN, 2016



Weiss-Penzias et al., 2019

Pros and Cons of Using Lichens as a Bioindicator of Atmospheric Deposition

Pros

- Less expensive than using instrumentation.
- Better measure of the net supply of the metal to terrestrial ecosystems than data from wet deposition measurements or model calculation.
- Hg contamination around point sources or “hot spots” can be easily depicted.
- Allows for the tracing of large-scale maps of Hg deposition and to locate “hot spots”.
- Validates the effectiveness of adopted environmental recovery procedures.

Cons

- The metal uptake and bioaccumulation depend on its bioavailability, the amount of deposition, pH, species, metabolic growth rate and surface area, which are all affected by climatic and environmental stress factors.
- Potential for poor distribution of lichen habitat across a broad landscape.
- Current need for standardization and optimization of biomonitoring procedures and for a more in-depth knowledge of the mechanisms and factors affecting Hg uptake and bioaccumulation.

Monitoring Need

- There is developing interest in the regulating communities in the West to limit public exposure to high levels of toxic methylmercury in fish.
- The California Water Board and Washington State Department of Ecology are developing TMDLs to limit mercury inputs to and accumulation within reservoirs.
- One of the principal, and unstudied, sources of mercury to reservoirs is atmospheric deposition (both to the reservoir and watershed).
- The mercury amounts in lichen can help to constrain atmospheric deposition, helping managers manage mercury inputs in an informed manner.

7.0 Water Quality (EN3)

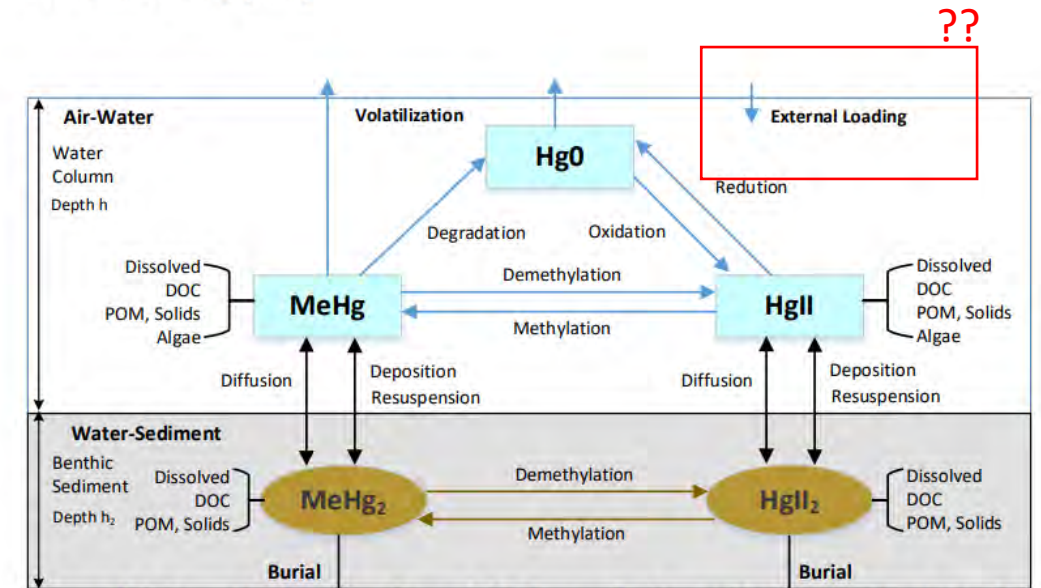
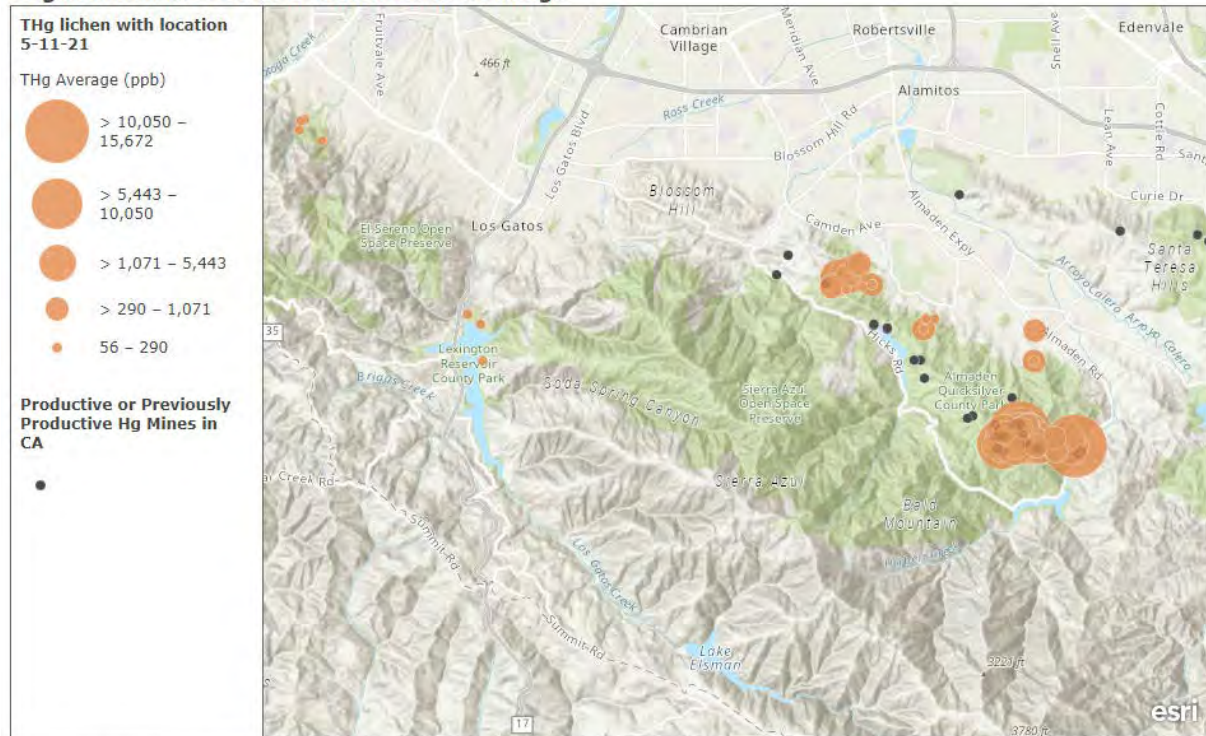


Figure 10. Example of research model linkages for improving understanding of mercury mobilization and transport in reservoirs. Image provided by Yong Lai, Reclamation.

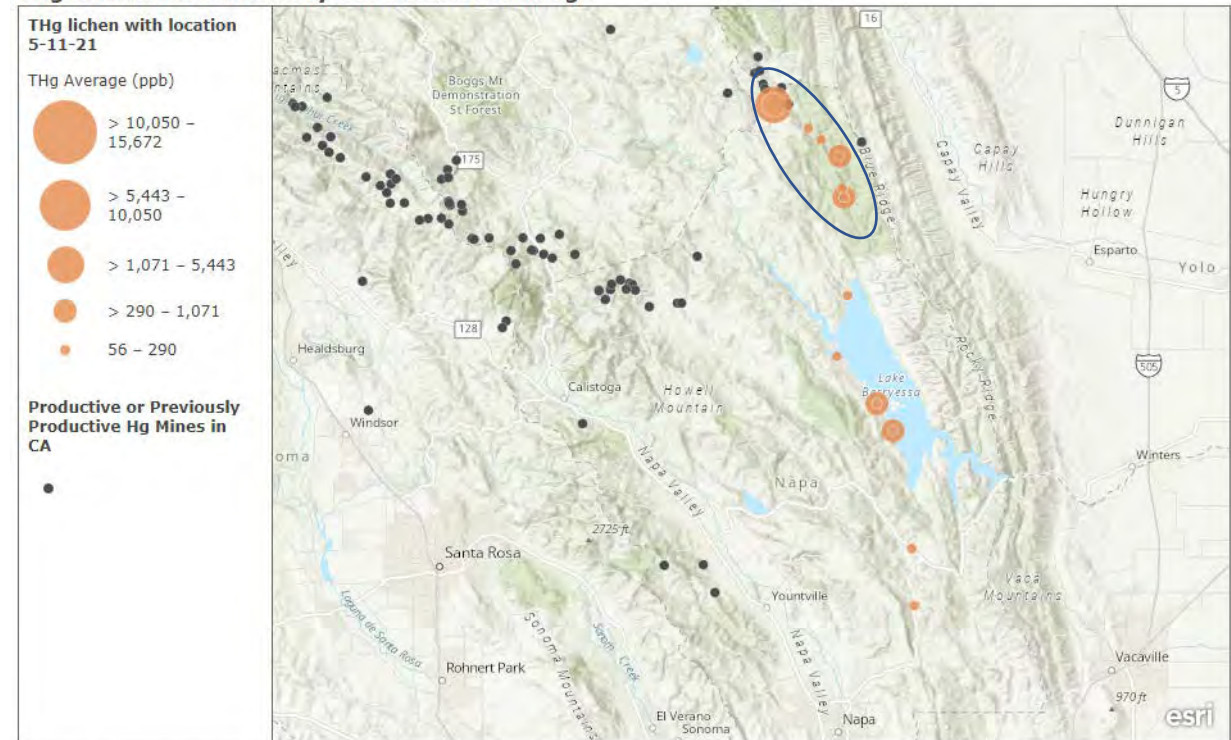
Mercury in Lichen at New Almaden and Lake Berryessa, 2020/2021

THg in lichen at New Almaden and surroundings



Esri, NASA, NGA, USGS | County of Santa Clara, Esri, HERE, Garmin, SafeGraph, METI/NASA, USGS, Bureau of Land Management, EPA, NPS, USDA

THg in lichen at Lake Berryessa and surroundings

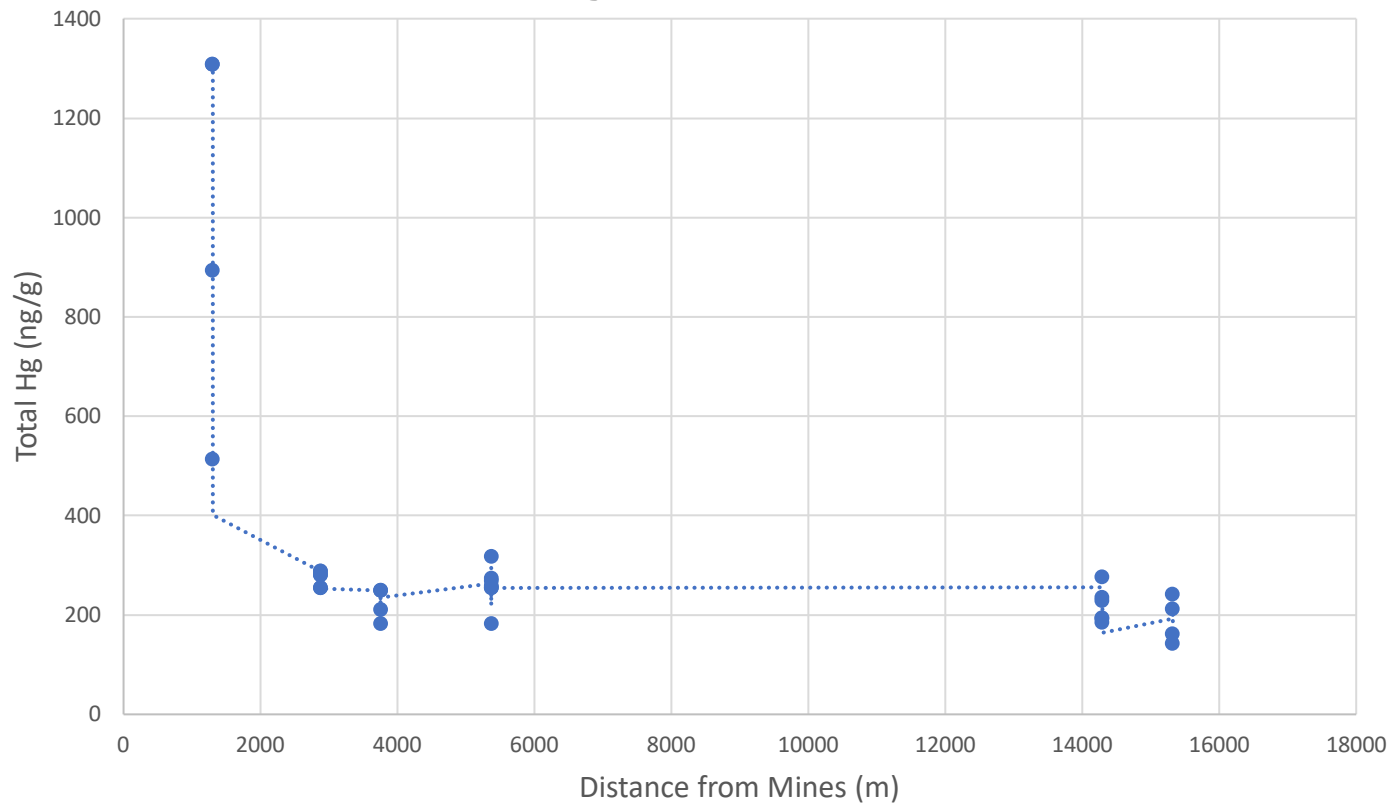


Esri, CGIAR, USGS | County of Napa, Yolo County, Esri, HERE, Garmin, SafeGraph, METI/NASA, USGS, Bureau of Land Management, EPA, NPS, USDA

Unpublished data

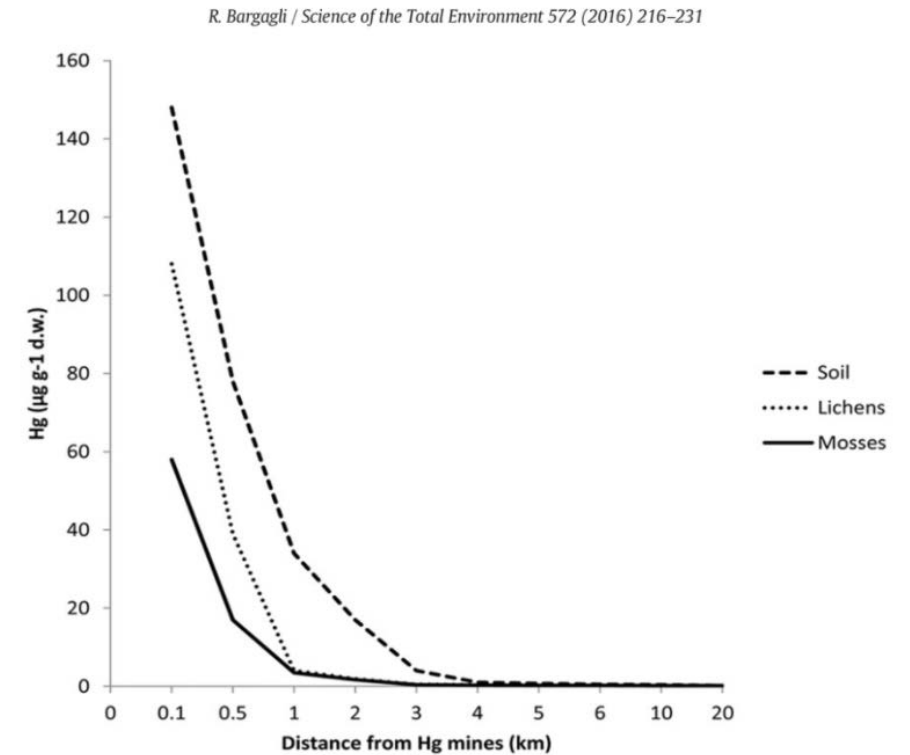
Total Hg in Lichen as a Function of Distance from Hg Mines

Knoxville to Lake Berryessa

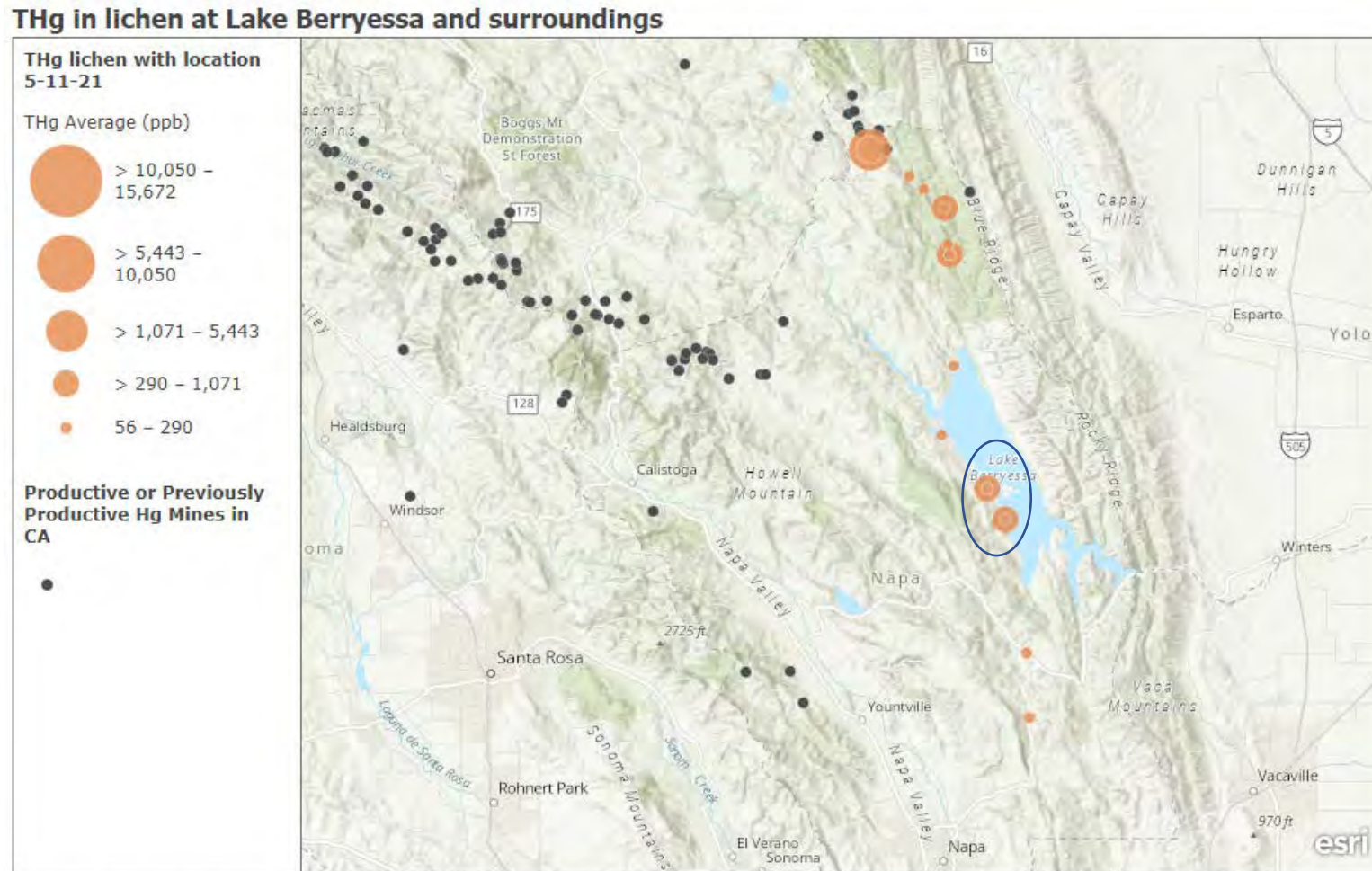


Unpublished data

Tuscany, Italy



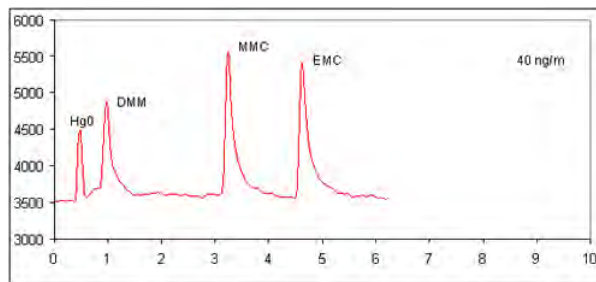
Elevated Lichen Hg at the Shore of Reservoir



- Fires?
- Mines?
- Free troposphere?

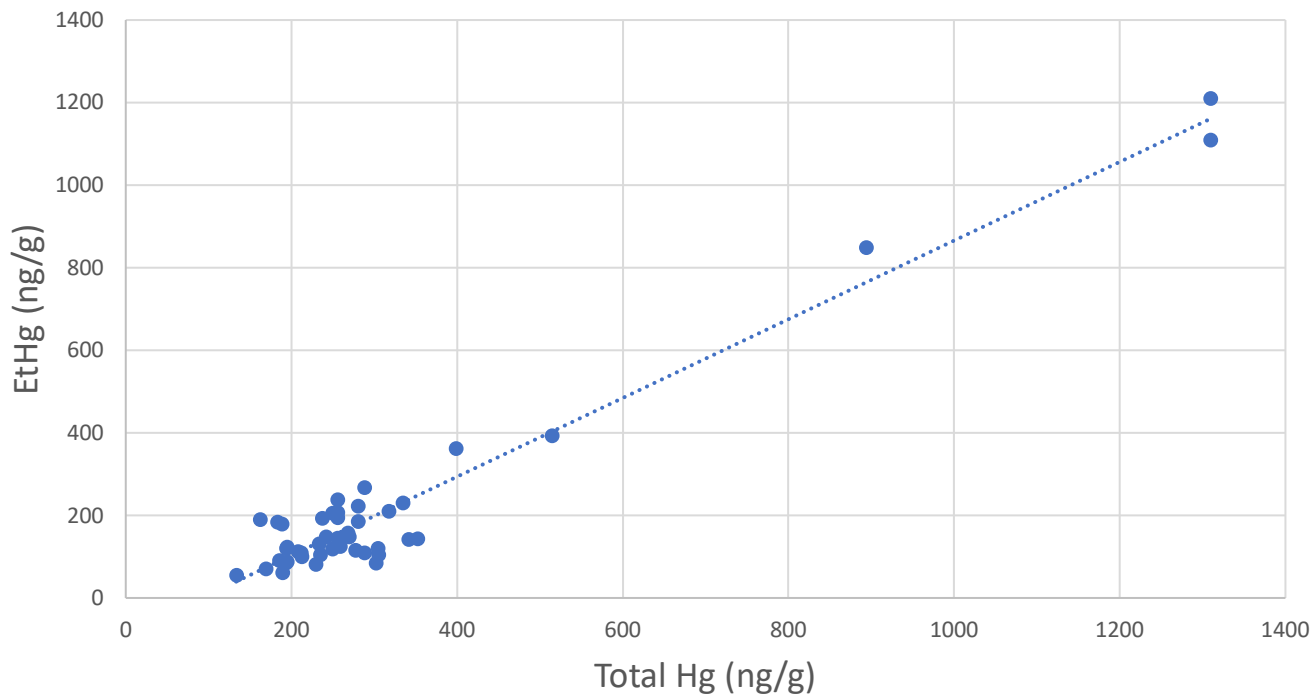
Speciated Hg vs. Total Hg in Lichen at Lake Berryessa

*EPA Method 1630
quantifies 4 Hg
species*



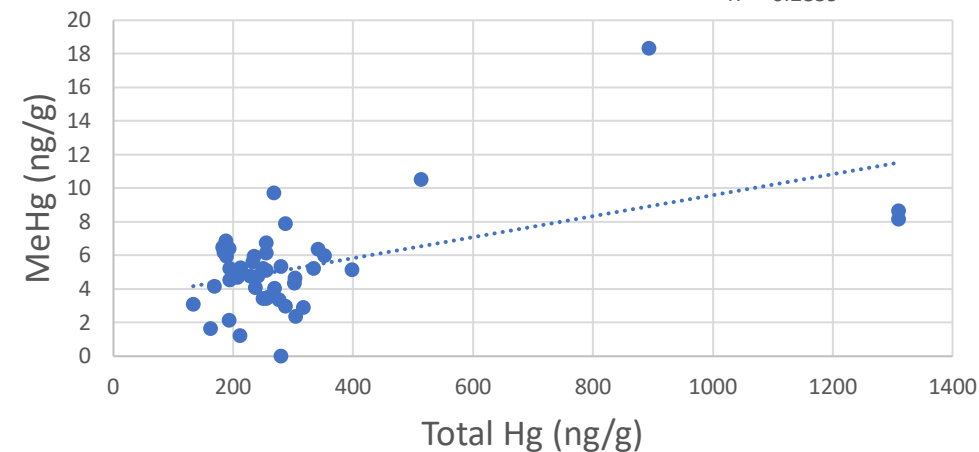
EtHg vs. total Hg

$$y = 0.9531x - 87.327$$
$$R^2 = 0.9441$$



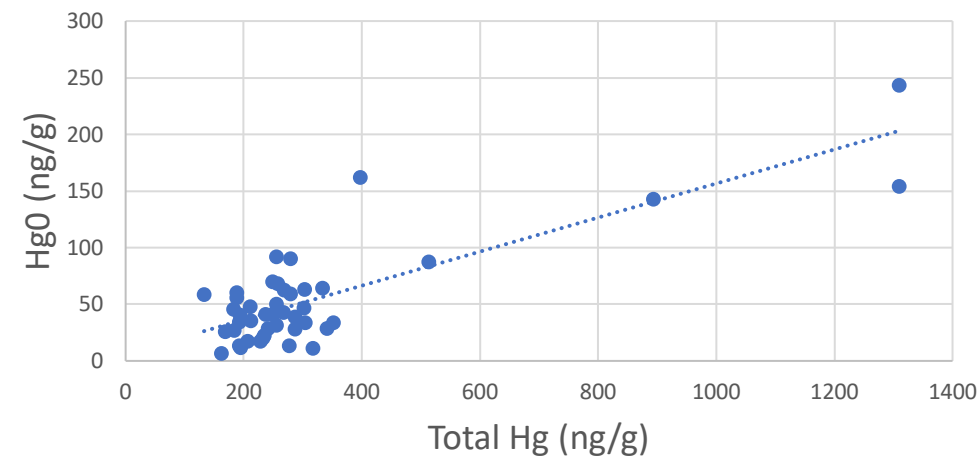
MeHg vs. total Hg

$$y = 0.0063x + 3.3137$$
$$R^2 = 0.2839$$

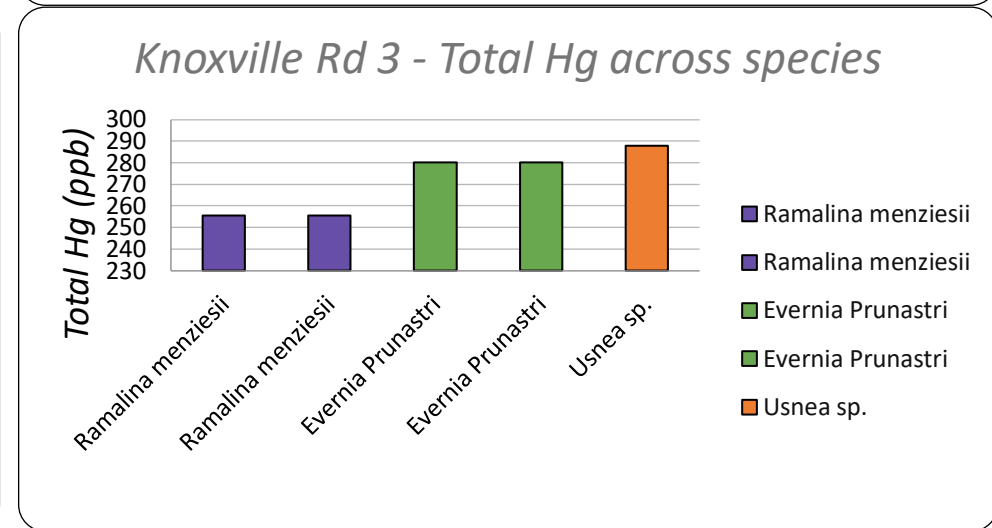
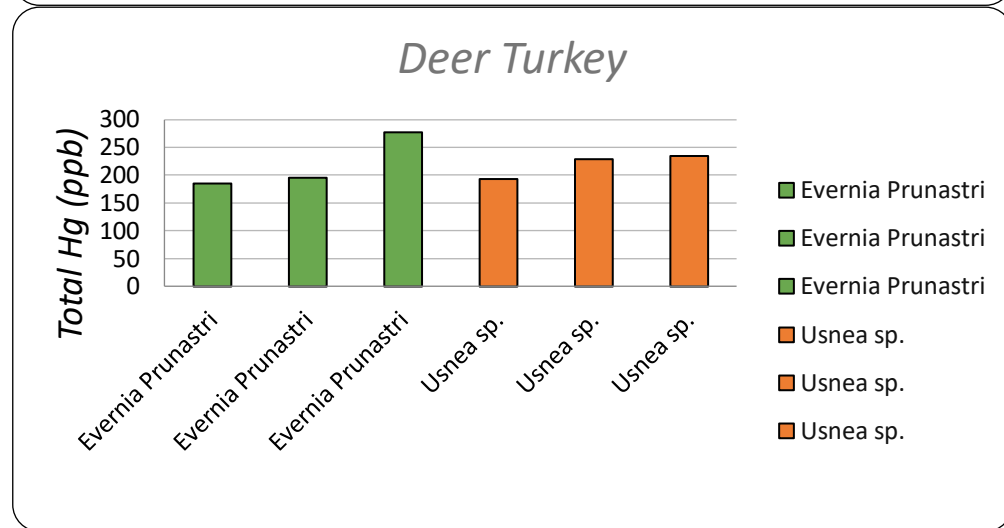
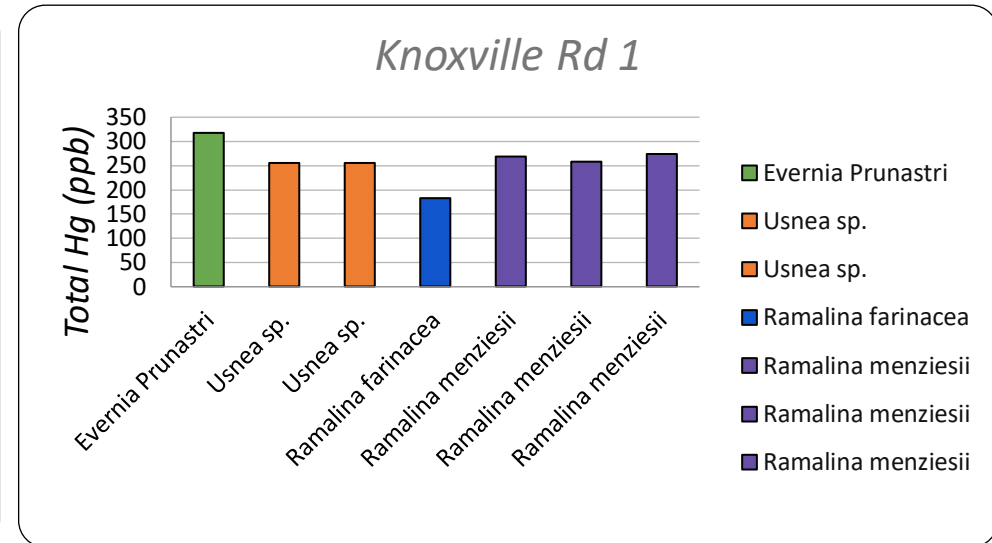
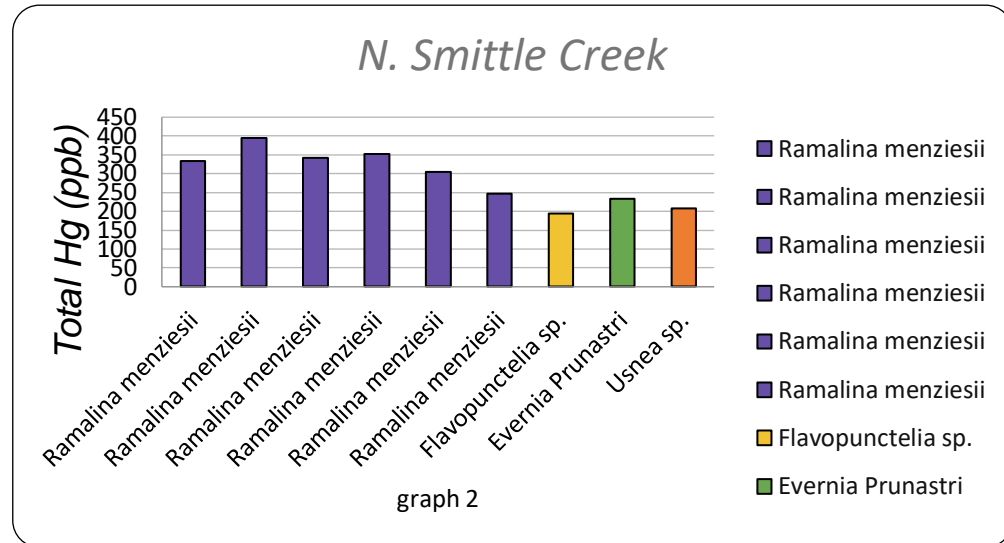


Hg0 vs. total Hg

$$y = 0.1503x + 6.2504$$
$$R^2 = 0.6609$$

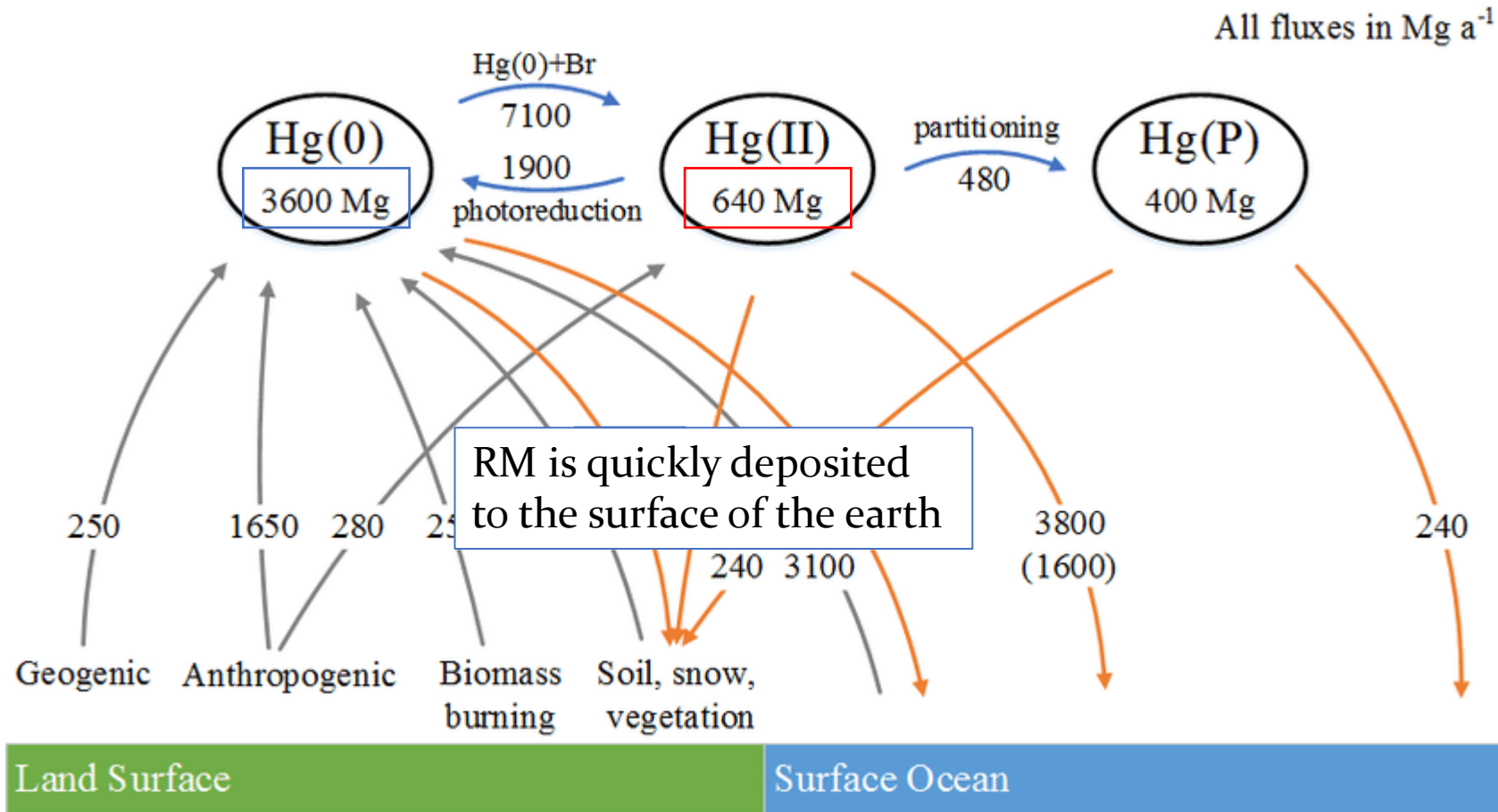


Total Hg as a Function of Species at Four Sites at Lake Berryessa



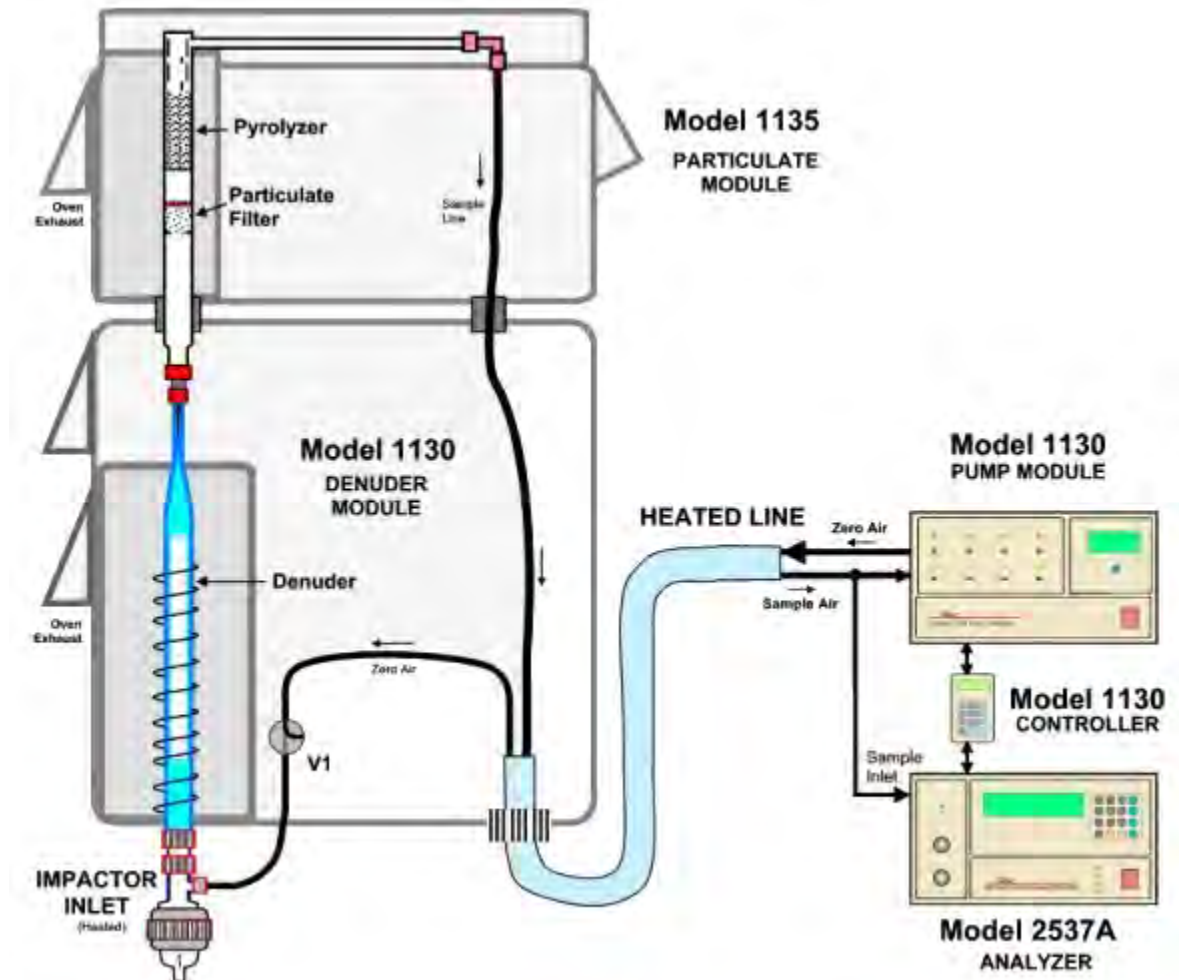
The Importance of Reactive Mercury in Atmospheric Deposition

Reactive Mercury (RM) = gaseous Hg(II) + Hg(P)

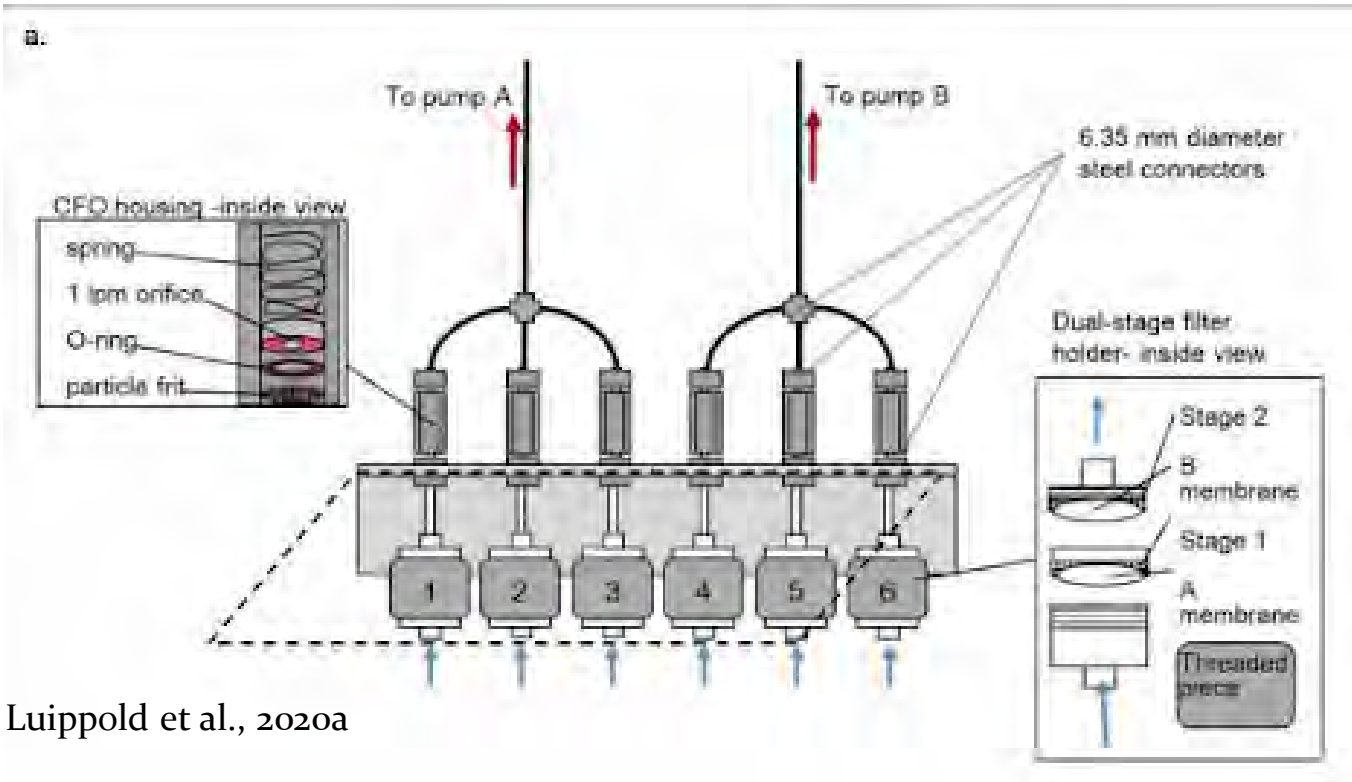


Traditional Instrumental Analysis of Reactive Mercury (Tekran 1130/1135/2537)

- Artefacts and biases due to water vapor and ozone
- Expensive system (~\$100k)
- Difficult to operate
- One instrument per site



New Instrumental Analysis: UNR-RMAS 2.0



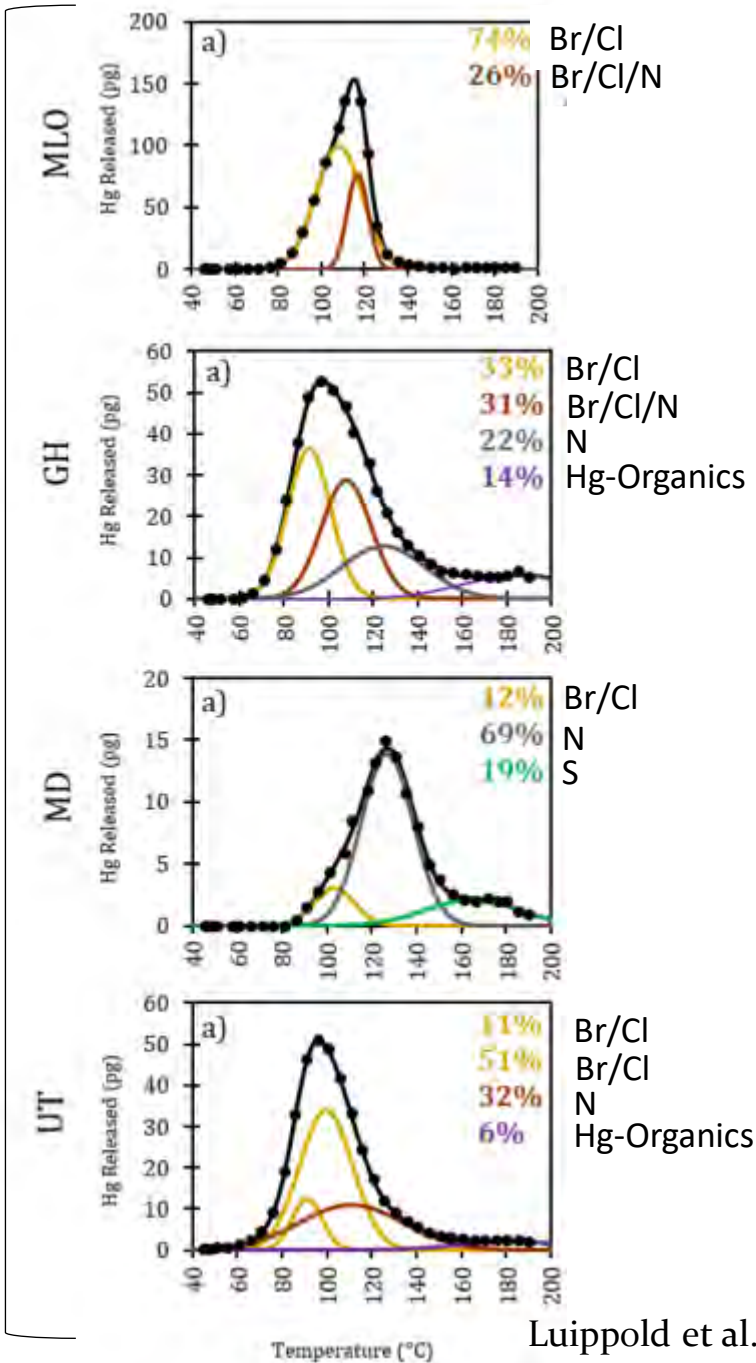
Luippold et al., 2020a

RM Concentration

Nylon and cation-exchange membranes are collected bi-weekly and analyzed for total Hg in the lab.

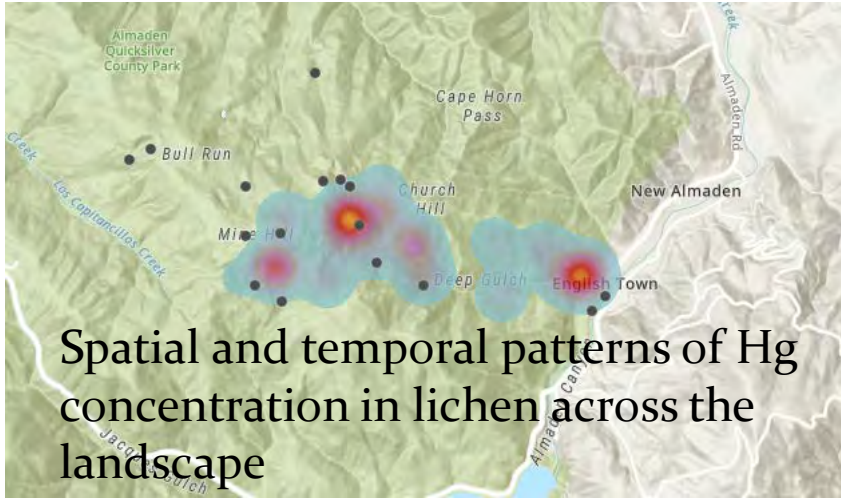
RM Speciation

Different RM compounds are desorbed at different temperatures.



Luippold et al., 2020b

New Idea #1: Calibrate Hg in lichen concentration maps, with atmospheric RM measurements and dry deposition modeling



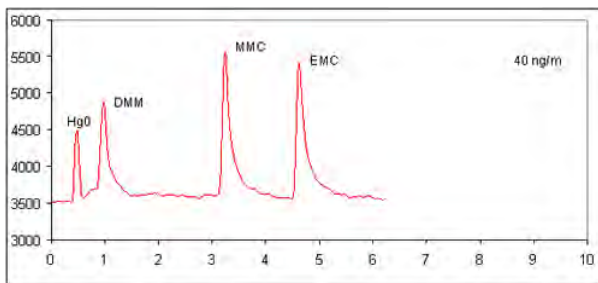
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RM concentration measurements at multiple locations + resistance modeling

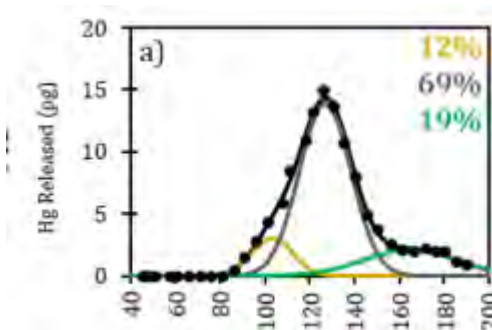


Landscape scale Hg deposition estimates

New Idea #2: Compare speciated Hg in lichen with atmospheric RM speciation to determine potential sources of Hg deposition

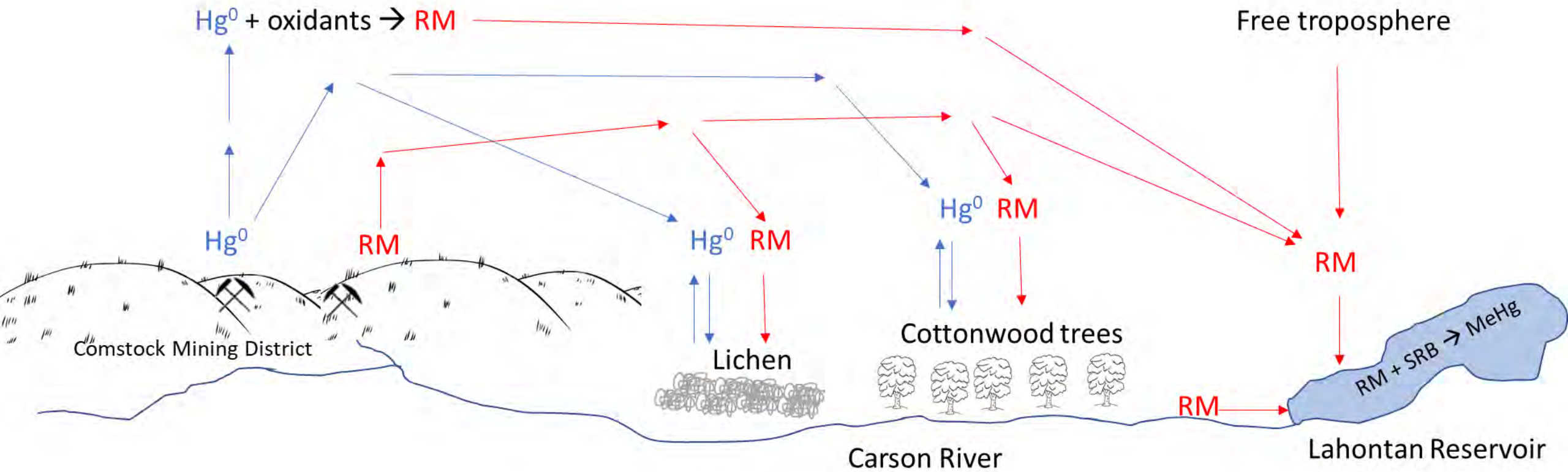


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Semi-quantitative assessment of sources of Hg deposition

UCSC/UNR/USBR Collaboration: Assessing the Susceptibility of Reservoirs and Watersheds to Pollution from Historical Mining Sites via the Atmosphere



- 1) Sample bioindicators in the watershed.
- 2) Make RM measurements and model dry deposition at two locations.
- 3) Combine RM and bioindicator Hg data to constrain atmospheric inputs to the reservoir.

Acknowledgements

- Daniel Deeds, Branch Chief, Environmental Monitoring and Assessment, US Bureau of Reclamation
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- Belle Zheng, Masters Student, UC Santa Cruz
- Tarabryn Grismer, Undergraduate Student, UC Santa Cruz



Tarabryn with the lichen picker at Lake Berryessa